

Liquid Nitriding of Stainless Steel 316L to Improve Fatigue Properties for Orthopedic Screws

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Abstract

Liquid nitrate is an important method used to improve mechanical properties, one of these properties is resistance to fatigue. The aim of this study was to improve the fatigue resistance of the stainless steel 316L. The rotational bending method was used with constant and variable stresses at different times of (1, 3, 5) hours and at (530, 630) °C. These tests were performed before and after nitration.

The results showed that the depth of the nitride layer was (0.21, 0.33, 0.45) mm, increasing with time nitriding when the temperature was 530 °C. While the depth of this layer at a temperature of 630 °C (0.26, 0.39, 0.5) mm with increasing time. As a result of these processes, a layer of solid chromium nitrides and other phases of iron nitride were formed on the outer surface. These layers helped to inhibit the growth of the cracks and their progress in addition to the generation of pressure stresses on the surface leading to obstructing the progress of the cracks.

This study showed that the fatigue resistance was directly proportional to the increase in nitrate time due to the increased depth of the hardened layer, but this resistance decreased when the temperature was 630 °C due to the formation of brittle phase with low resistance.

Keywords: Liquid Nitriding, SS316L, Fatigue, Brinell Hardness.

Introduction:

Austenitic stainless steels stainless 316L are used in different industrial extremely corrosive and biomedical applications. These soft steels have low hardness and wear protection resistance. Accordingly, their application is limited. Surface Engineering applied by many researchers by adjusting material's surface to enhance surface properties to improve the surface contact with the atmosphere. This enhanced their erosion protection and strength in addition to their ability to provide appropriate properties at economic cost on a production scale [1,3].

In the past decades, high nitrogen steels (HNS) have been regarded as substitutes for conventional austenitic stainless steels because of their superior

mechanical and corrosion properties. However, the main limitation to their wider application is their expensive production processes. As an alternative, high temperature solution nitriding has been applied to produce HNS from three commercially available stainless-steel grades (AISI 304L, AISI 316 and EN 1.4369) [4]. As for the powder metallurgy process, the dissolution of nitrogen occurs in the solid state, where the solubility is higher [5].

Since the middle of the 80s, low temperature thermochemical treatments have been developed for surface hardening of austenitic stainless steels, including gas carburizing and plasma nitriding. These processes can induce formation of a precipitate-free interstitially supersaturated metastable expanded austenite, also known as S-phase, having superior hardness and improved wear resistance, while maintaining corrosion resistance [6].

Liquid nitriding of austenite stainless 321 steel was conducted at low temperature of 430 °C, using a type of a complex chemical heat-treatment; and the properties of the nitride surface were evaluated. Experimental results revealed that a modified layer was formed on the surface with the thickness ranging from 2 to 30 μm varying with changing treatment time. When the stainless steel subjected to the advanced liquid nitriding less than 8 h at 430 °C, the main phase of the nitride coating layer was the S phase generally. When the treatment time prolonged up to 16 h, S phase formed and partially transformed to CrN subsequently; and then the fine secondary CrN phase precipitated [7].

Compared to conventional gas nitriding and ion nitriding, the liquid nitriding treatment (salt bath nitriding) was regarded as an effective, low-cost method with many advantages, such as low treatment temperature, short treatment time, high degree of shape and dimensional stabilities, and reproducibility [8].

Nitriding Enhancements gained through plasma nitriding without affecting their corrosion resistance. Where, this method applied in temperature range from (350-570) °C for two hours. The different types of nitrides which formed improved the microhardness, which improved microwear behavior and the corrosion resistance [9].

Liquid nitriding applied in a range of temperatures between (400 - 670) °C for carbon steel C110. The nitride layer depth increased significantly with increasing the treating temperature. The liquid nitriding effectively improved the surface hardness and dependently tensile strength also. After liquid nitriding, the absorption energy of the treated sample decreased. In addition to elongation decreasing. Despite of treatment temperature, the liquid nitriding can improve the corrosion [10].

The fatigue strength was improved by the plasma nitriding and the shape of the S-N curves was a asymptote shape with a fatigue limit up to N = 108 cycles. The fatigue cracks initiated from specimen surface in all specimen. However, the improvement of fatigue strength by plasma nitriding and the influence of nitriding time on the fatigue strength was small because the nitriding layer was quite thin, it is believed that the nitride layer could not affect significantly the fatigue strength. [11] meanwhile the fatigue behavior of EN8 steel under different treatments such as coating, nitriding, induction hardening and combined nitriding and induction hardening. The EN8 steel was exposed to Nitriding in cyanide salt bath with 560°C. Results showed that coating on nitride specimens produces high fatigue resistance [12].

According to above review it could be concluded that the nitriding was conducted with different ways to investigate its effect on different mechanical properties. Meanwhile, only plasma nitriding has been used to study it effect on fatigue property. In this work liquids nitriding will be applied to investigate the effect of fatigue properties.

Experimental Part:

1. Materials:

The alloy used in this study is 316L steel with chemical composition shown in Table (1). These samples received as a rod with a diameter of 10 mm. The Mechanical Properties of these samples also illustrated in Table (2).

Table (1) The Chemical Composition of 316 Stainless steel rods as received from Provider according to data sheets that supplied in the provider website

C	Si	Mn	P	S	Cr	Mo	Ni
0.03	0.73	1.9	0.044	0.03	16.8	2.6	11.3

Table (2) mechanical properties of 316 Stainless steel rods as received from Provider according to data sheets that supplied in the provider website.

UTS (MPa)	0.2% YS (MPa)	Elongation %	HB
480	175	42	210

2. Experimental Procedure:

The Stainless-Steel rods operated on a CNC lathe to obtain high accuracy fatigue samples. The sample surfaces were then smoothed to eliminate the stress concentration areas. Standard E-8 dimensions of the sample used in rotational bending tests. As shown in Figure (1).

After preparing the fatigue samples, all samples were thermally treated by quenching treatment to obtain a solid heart and a hard surface. This heat treatment included heating the samples to a temperature of 980 °C and fixed at this degree for one hour after that quenched in oil.

Tempering Treatment applied at a temperature of 600 °C for one hour and then air cooled. This treatment procedure explained in Figure (2)

Liquid Nitriding was carried out in molten salt baths. Before nitriding the samples were placed in oven at a temperature of 120°C for 30 minutes for the purpose of release the dirt and oil. The samples were then cleaned in special containers and placed in the furnace for the nitrate, after the furnace was heated to 530 C° for 1 hour. The above treatment was repeated on other samples of the same steel but for 3 hours and then for 5 hours. The samples were removed from the furnace and cooling in air. The nitride thermal treatment was then performed on other samples and at the same previous time periods but at a temperature of 630°C.

The rotating bending fatigue tester Hi-Tech shown in Figure (3) have been used to investigate the fatigue life time.

Grinding and polishing operations were performed to all the samples in order to investigate the macro hardness. Brinell Macro-Hardness measurements have been done using prufmas chinen 3030 rating device. Five readings were taken and the average of these five readings was concluded, to measure microscopic hardness at successive depths of the sample surface.

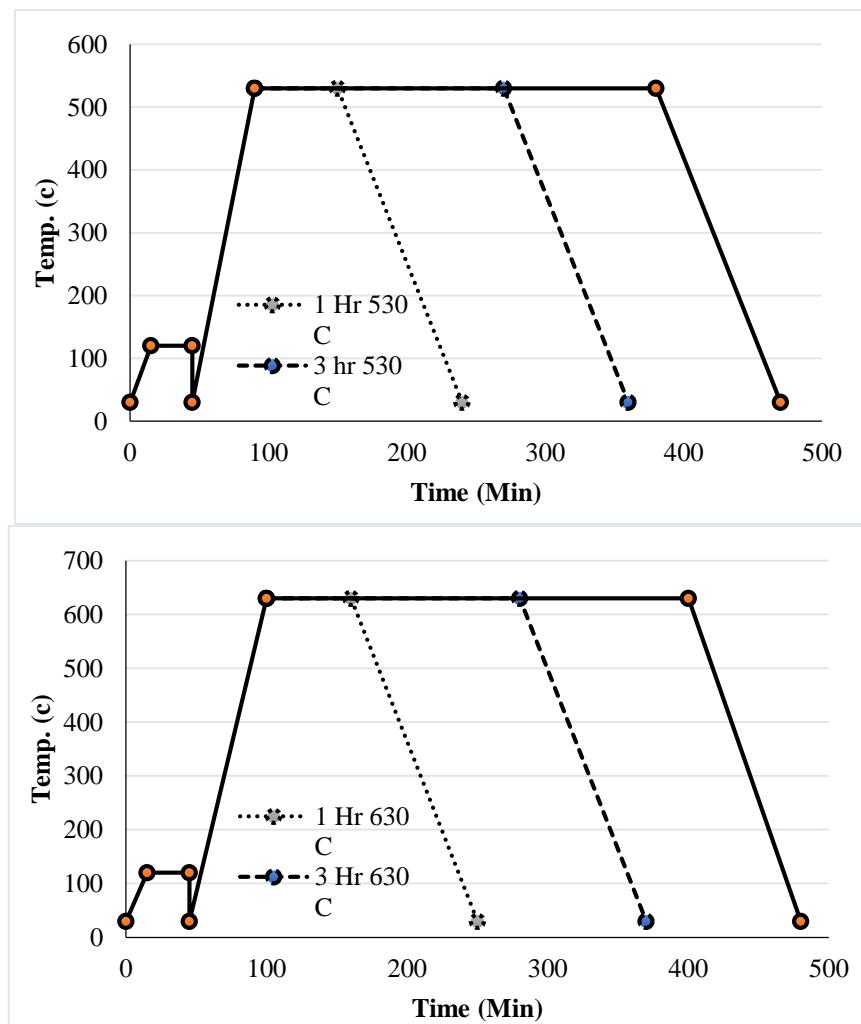
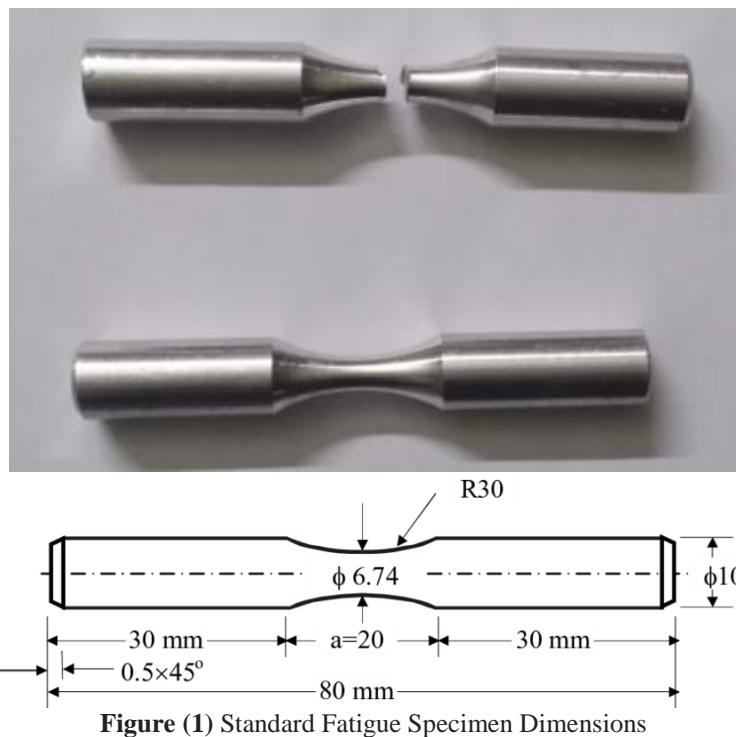


Figure (2) Heat Treatment Plans of Nitriding with three times (1,3 and 5 Hrs) with two different Temperatures (530 and 630) C°



Figure (3) Fatigue test device

Results

The hardness – distance plots shown in Figure (4.A and 4.B) showed that the hardness increased with nitriding time meanwhile the depth of layer remains the same. This showed that the nitrided was slightly increased with temperature and not effected with time.

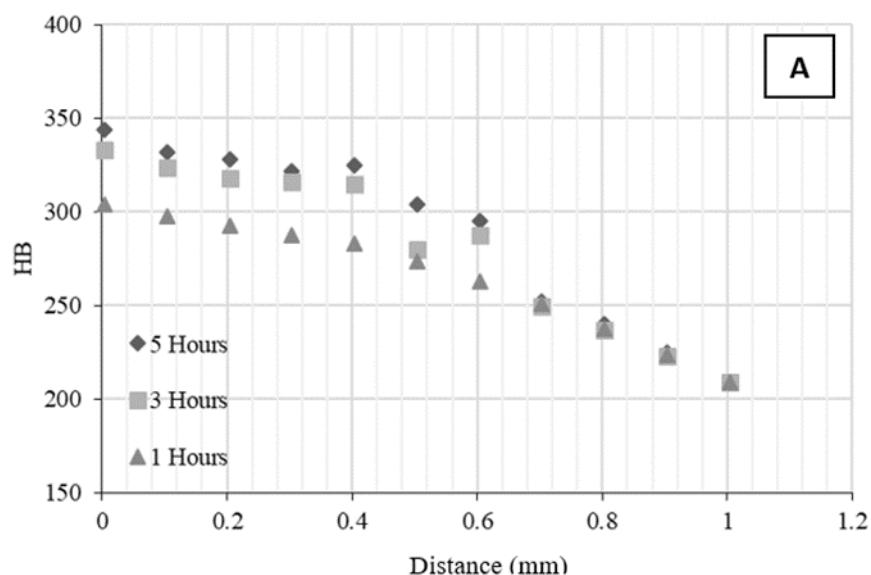
The two sets of samples that have been nitrided with (530 and 630) C° with the three times have been investigated by Brinell Hardness along 1 mm from outer surface towards the center of samples. The reads that illustrated in Table (3) shows the hardness values variation with distance.

According to this result, it was clearly seen that the fatigue resistance also increased with hardness values. As shown in Figure (5.A 5. B and 5.C). and the variation in fatigue limit stress was more clearly increased with treatment time as shown in Table (4). These results compared with annealed

Samples to indicated the effect of nitriding far from annealing process that have been done. The depth of the hardened layer did not change according to the drawing, but the value of the fatigue increased with time due to the increase in the value of the hardness and not the depth and thus affect the initiation and propagation of crack while the depth affects the progress of this crack.

Variable load: Study of the effect of the change in the capacity of the fatigue stress at the age of samples of different nitrate time.

Figure (6) showed the loading program (low-high, high-low). The nitrification process showed that the samples were older than four, five and seven times respectively at recessive times (1,3,5) (Low to high). At high loading, the increase in ages was three, four, and six times respectively at receding times (1,3,5). This is because the cracks grow and appear faster at loading (high-low).



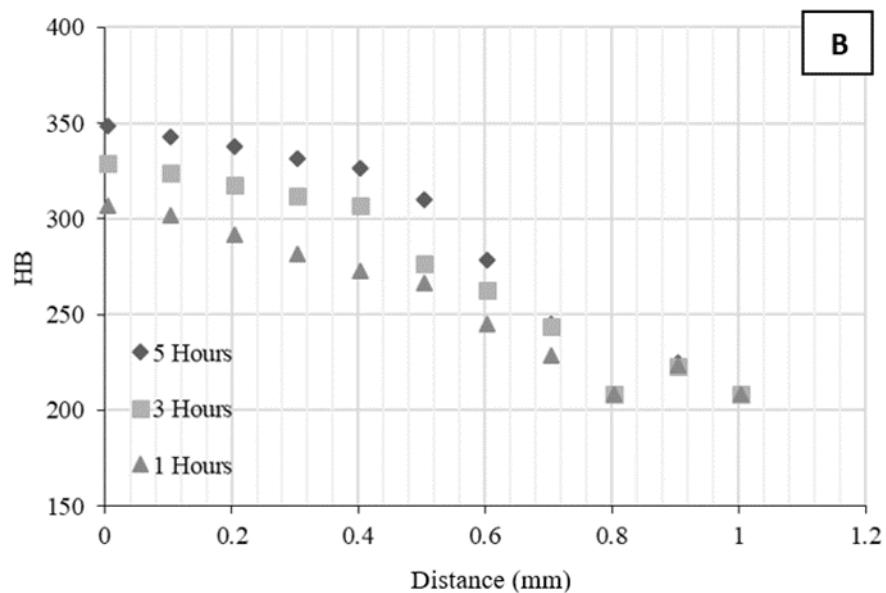
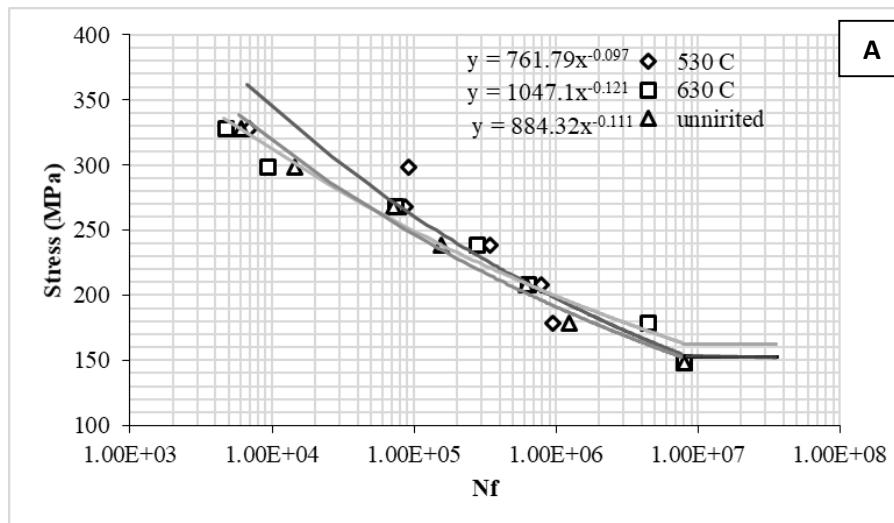


Figure (4) The relationship between the brinell hardness and distance from the surface of the Nitrided Samples for 1,3 and 5 hours at **A:** 530°C and **B:** 630°C

Table (3) Brinell hardness Values along the distance from surface to core for nitrate samples at 530 and 630 C° for 1,3 and 5 Hours

Distance (mm)	HB for nitride Samples at 630°C			HB for nitride Samples at 530°C		
	1 Hour	3 Hours	5 Hours	1 Hour	3 Hours	5 Hours
0	305	334	345	308	330	350
0.1	299	325	333	303	325	344
0.2	294	319	329	293	319	339
0.3	289	317	323	283	313	333
0.4	284	316	326	274	308	328
0.5	275	281	305	268	278	311
0.6	264	289	296	246	264	280
0.7	252	251	253	230	245	246
0.8	239	238	241	210	210	210
0.9	225	224	226	225	224	226
1	210	210	210	210	210	210



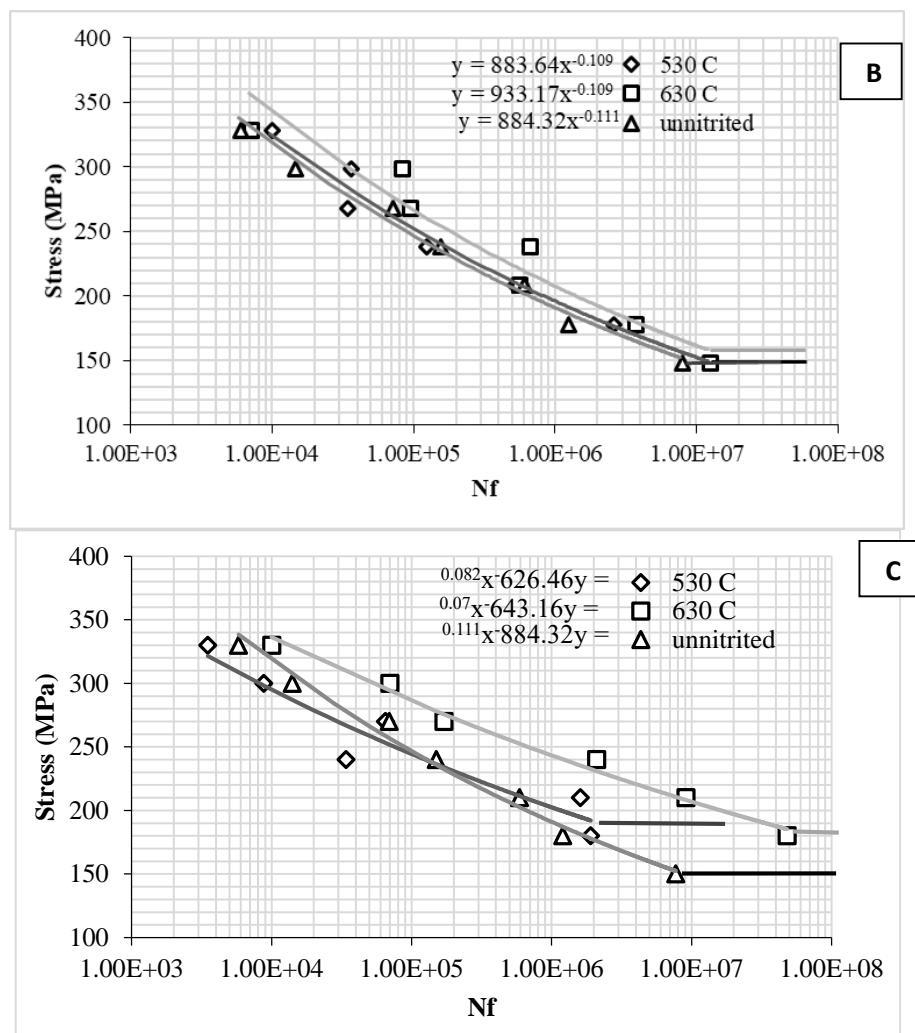


Figure (5) The S-N Curves for the un-nitrited Samples with Nitrited Samples at 530C⁰ and 630C⁰ for **A:**1 Hour, **B:**3 Hours and **C:**5 Hours

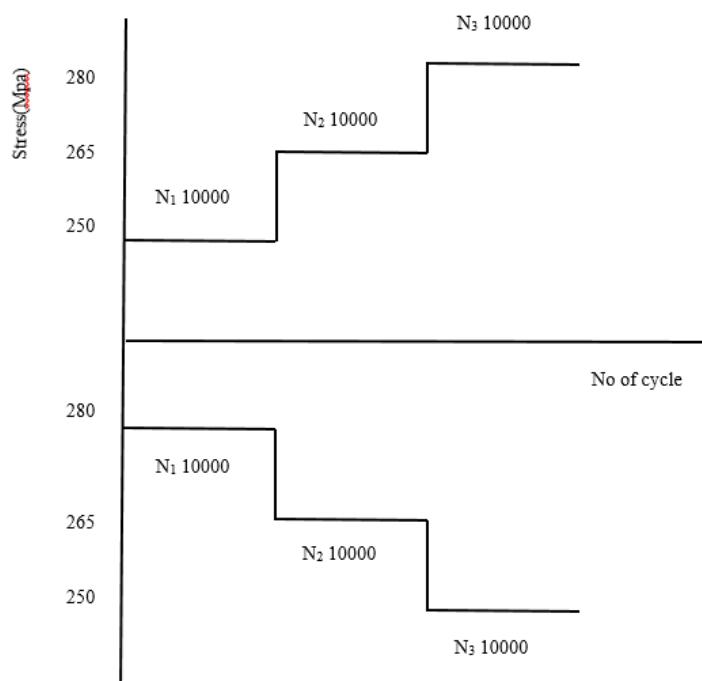


Figure (6): Load program (low-high) at (250-280) Mpa and (high-low) at (280-250) Mpa at Ns = 10000

Table (4) The Relationship Between Fatigue Life and Nitriding Time at Different Temperatures

Nitriding Time (Hours)	Fatigue Limit (MPa)	
	630°C	530°C
1	156	160
3	160	165
5	185	190

Table (5) The results of examination of the accumulated damage in the samples that were nitriding at different time at 530C°

Loading Sequence	Stress (MPa)	N _s	N _f
Without Nitriding			
L - H	250-280	10000	195443
H - L	280-250	10000	244600
Nitriding For 1 Hour			
L - H	250-280	10000	295300
H - L	280-250	10000	414542
Nitriding for 3 Hours			
L - H	250-280	10000	495830
H - L	280-250	10000	799901
Nitriding for 5 Hours			
L - H	250-280	10000	891223
H - L	280-250	10000	1666570

Discussion:

This study showed that the depth of the nitride layer increases with the increased duration of the nitride process at constant temperature. The depth of the nitride layer was 0.21 mm when the nitrate time was 1 hr, 0.33 mm at 3 hr and 0.45 mm at 5 hr when the temperature was 530 C°.

Note that the higher depth of the nitride layer, the faster spread of the hydrogen atoms, making it difficult to reach deep ranges of nitrate. When the nitride process was performed at 630 C° and the same previous time intervals 1,3, and 5 hr, the depth of the nitride layer was found to be 0.26, 0.39, 0.5 mm respectively. When comparing the results between the two cases it was observed that the depth of the nitride layer was greater when the process at 630 C°, this could be due to the presence of some elemental alloys such as phosphorus and sulfur led to the formation of brittle phases, which the decrease of hardness values as a approved by previse studies [4,7,8].

The primary purpose of thermal treatment prior to the nitride process was to obtain a solid heart and a hard surface and this condition was not affected by nitrate processes. The results of this work showed that the best fatigue life and hardness values could be resulted with higher temperatures at which the nitrate processes are (650) C° and this depends on the method of nitride that will be conducted whether gas or liquid or solid [10,12].

The fatigue limit increased after the nitrate process and the value of this increase was proportional to nitrate time. This increase was due to the formation

of hard chromium nitrides on the surface of the steel which will work agenst crack initiation and propagation.

Finally, when variable cyclic stresses are applied, the age of fatigue is higher at loading (high - low) compared to the loading (low - high), when these samples are with nitride or without it. This is due to the presence of a ductile zone in the top of crack that prevents these cracks from advancing.

Conclusions:

- 1- The age of the fatigue is increased as the nitrate process increases, the nitrate process at 530C° is better than 630C°
- 2- Increase the depth of the hard layer with increasing time of the nitride process.
- 3- Increasing the temperature to more than 650 C° leads to a reduction in the threshold of fatigue. this depends on the method of nitride that will be conducted whether gas or liquid or solid [10,12].
- 4- The age of fatigue is greater when loading (high - low) compared to loading (low - high), because the presence of a ductile zone in the top of crack that prevents these cracks from advancing.

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النيتروجين السائل من الفولاذ المقاوم للصدأ L316 لتحسين خاصية الإلعاياء لمسامير العظام

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قسم

الخلاصة:

ان طريقة النتردة السائلة هي من الطرق المهمة والمستخدمة في تحسين الخواص الميكانيكية ومن هذه الخواص مقاومة الكل. الهدف من هذا البحث هو تحسين مقاومة الكلل للفولاذ المقاوم للصدأ الاوستينيتي L316, تم استخدام طريقة الانحناء الدوراني مع تسليط اجهادات ثابتة ومتغيرة وبأوقات مختلفة وهي 5,3,1 درجة مئوية وعند درجات حرارة 630,530 درجة مئوية وقد اجريت هذه الاختبارات قبل وبعد عملية النتردة.

بينت النتائج ان عمق الطبقة المصلدة كانت 0.45,0.33,0.21 ملم تصاعديا مع الزمن عندما كانت درجة الحرارة 530 درجة مئوية. بينما بلغ عمق هذه الطبقة عند درجة حرارة 630 درجة مئوية 0.5,0.39,0.26 ملم مع زيادة الزمن، ونتيجة لهذه العمليات تكونت على السطح الخارجي طبقة من نتريدات الكروم الصلدة واطوار اخرى من نترید الحديد وقد ساعدت هذه الطبقات على اعاقة نمو الشفوق وتقدمها اضافة الى توليد اجهادات ضغط على السطح تؤدي الى عرقلة نقدم الشفوق.

تبين من هذه الدراسة ان مقاومة الكلل تتاسب طرديا مع زيادة زمن عملية النتردة بسبب زيادة عمق الطبقة المصلدة، لكن هذه المقاومة انخفضت عندما كانت درجة الحرارة 630 درجة مئوية بسبب تكون اطوار هشة ذات مقاومة ضعيفة.

تعتبر طريقة النتردة السائلة هي من الطرق المستخدمة في تحسين الخواص الميكانيكية خلال الدراسة تم تحسين مقاومة الكلل والتي هي احدى الخواص الميكانيكية المهمة للفولاذ المقاوم للصدأ الاوستينيتي L316, حيث استخدام طريقة الانحناء الدوراني مع تسليط اجهادات ثابتة ومتغيرة وبأوقات مختلفة وهي 5,3,1 درجة مئوية وعند درجات حرارة 630,530 درجة مئوية وقد اجريت هذه الاختبارات قبل وبعد عملية النتردة.

من خلال النتائج تبين ان عمق الطبقة المصلدة كانت 0.45,0.33,0.21 ملم تصاعديا مع الزمن عندما كانت درجة الحرارة 530 درجة مئوية. بينما بلغ عمق هذه الطبقة عند درجة حرارة 630 درجة مئوية 0.5,0.39,0.26 ملم مع زيادة الزمن، ونتيجة لهذه العمليات تكونت على السطح الخارجي طبقة من نتريدات الكروم الصلدة واطوار اخرى من نترید الحديد وقد ساعدت هذه الطبقات على اعاقة نمو الشفوق وتقدمها اضافة الى توليد اجهادات ضغط على السطح تؤدي الى عرقلة نقدم الشفوق.

اثبت من خلال هذه الدراسة ان مقاومة الكلل تتاسب طرديا مع زيادة زمن عملية النتردة بسبب زيادة عمق الطبقة المصلدة، لكن هذه المقاومة انخفضت عندما كانت درجة الحرارة 630 درجة مئوية بسبب تكون اطوار هشة ذات مقاومة ضعيفة.